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Variation in the Deposition Velocity of a Reactive Gas

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## Variation in the Deposition Velocity of a Reactive Gas

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### Abstract

The flux of a gas to the surface arising from absorption (or other chemical and physical processes) is commonly represented by the product of the concentration of the gas at some height  $z$  and a velocity, called the deposition velocity,  $v_d$ . Since the concentration changes with height, the deposition velocity must also change. Furthermore since the flux of the gas to the surface and its uptake there depends on the surface fluxes of momentum and heat,  $v_d$  must also be made a function of these variables too. If we follow Chamberlain (1966) and represent the extra surface resistance in terms of  $B$ , the Stanton number of Owen and Thomson (1963), which Thom (1972) expressed in terms of the surface friction velocity  $u_*$ , then  $v_d$  can be determined as a function of wind speed and temperature gradient.

### The Analysis

For any rough surface, there is less resistance to the air to surface transfer of momentum than to the transfer of gaseous material or heat because momentum can be taken up through the action of pressure forces whereas the others depend entirely on molecular conduction. The difference in the resistances has been expressed in terms of the dimensionless reciprocal Stanton number  $B^{-1}$  introduced by Owen and Thomson (1963), defined by :

$$v_d(z) = \frac{u_*^2}{u(z)} \left[ 1 + \frac{u_*}{u(z)} B^{-1} \right]^{-1} \quad (1)$$

Experiments carried out by Chamberlain (1966), Rider (1954) and Thom (1972) strongly suggest that  $B$  varies very little over a wide range of vegetated surface-types.  $B$  does vary, however, with the friction velocity  $u_*$ . Thom finds best agreement with a relationship of the form:

$$B^{-1} = a u_*^{1/3} \quad (2)$$

in which  $a$  is purely a function of the character of the gas being deposited. For example  $a(\text{Thorium B}) / a(\text{water vapour}) = 1.6$ , according to Chamberlain (1966). The results presented in this paper are for  $a = 10$  (when  $u_*$  is in  $\text{m s}^{-1}$ ), which is appropriate for Thorium B, a fairly reactive gas used by Chamberlain. Figure 1 suggests that for other materials in which the coefficient  $a$  might range from 0 to 40 the value of  $v_d$  would change from roughly  $2v_d(\text{ThB})$  to  $\frac{1}{2}v_d(\text{ThB})$ .

Although  $B^{-1}$  appears to be very insensitive to the surface, typified by the roughness length  $z_0$ , the deposition velocity  $v_d(z)$  depends much more sensitively on  $z_0$  through the dependence of  $u_*$  and  $u(z)$  on  $z_0$ , by which  $v_d$  is expressed (eqn (1)).



This dependence is shown in Figure 2 for a given geostrophic wind ( $G = 10 \text{ m s}^{-1}$ ). The Figure also shows how relatively insensitive  $v_d$  at a height of 1 metre is to the surface sensible heat flux  $H$ , at least on the unstable side when  $H \geq 0$ .

To calculate  $v_d$ , for a given value of the parameter  $a$  and a specified  $z_0$ , values of  $u_*$  and  $u(z)$  have to be calculated. It will be assumed that  $u(2)$ , the wind speed at a height of 2 metres, is available together with a temperature difference between 1 metre and 8 metres.

The following Monin-Obukhov similarity relations are assumed to be valid within the surface layer of the boundary layer:

$$\frac{du}{dz} = \frac{u_*}{kz} \phi_m\left(\frac{z}{|L|}\right) \quad (3)$$

$$\text{where } L = - \frac{\rho c_p T u_*^3}{kgH}$$

$$H = \text{the surface sensible heat flux} = \rho c_p u_* T_* \quad (\text{Wm}^{-2})$$

$$T = \text{temperature} \quad (^{\circ}\text{K})$$

$$\rho = \text{air density}$$

$$c_p = \text{specific heat at constant pressure}$$

$$g = \text{acceleration due to gravity}$$

$$k = \text{von Karman's constant, taken to be } 0.4$$

Ignoring small changes in temperature and their effect on  $\rho c_p$  ( $\rho c_p \approx 1305 - 4.3(T - 273)$ ) we may write

$$L \approx 29.358 \frac{u_*^2}{T_*} \quad (4)$$

Following Dyer and Bradley (1982) on the unstable side ( $L$  negative):

$$\phi_m = \left(1 + 28 \frac{z}{|L|}\right)^{-1/4} \quad \text{for } 0 < \frac{z}{|L|} < 4 \quad (5a)$$

equation (3) may be integrated to give:

$$u(z) = \frac{u_*}{k} \left\{ \ln \left[ \frac{s-1}{s+1} \cdot \frac{s_0+1}{s_0-1} \right] + 2 \tan^{-1} s - 2 \tan^{-1} s_0 \right\} \quad (6a)$$

$$\text{where } s = \left(1 + 28 \frac{z}{|L|}\right)^{1/4}$$

$$\text{and } s_0 = \left(1 + 28 \frac{z_0}{|L|}\right)^{1/4}$$

On the stable side we take Webb's (1970) form:

$$\phi_m = \left(1 + 5.2 \frac{z}{L}\right) \quad \text{for } 0 < \frac{z}{L} < 1 \quad (5b)$$

which yields

$$u(z) = \frac{u_*}{k} \left\{ \ln \frac{z}{z_0} + \frac{5.2}{L} (z - z_0) \right\} \quad (6b)$$

$$\text{Similarly } \frac{dT}{dz} = \frac{T_*}{k} \phi_H \left( \frac{z}{|L|} \right) \quad (7)$$

On the unstable side, we again use Dyer and Bradley's results:

$$\phi_H \left( \frac{z}{|L|} \right) = \left(1 + 14 \frac{z}{|L|}\right)^{-1/2} \quad \text{for } 0 < \frac{z}{|L|} < 4 \quad (8a)$$

whereas on the stable side, following Webb:

$$\phi_H \left( \frac{z}{L} \right) = \left(1 + 5.2 \frac{z}{L}\right) \quad \text{for } 0 < \frac{z}{L} < 1 \quad (8b)$$



The temperature difference between 1 and 8 metres is therefore

$$\Delta\theta = \frac{T_*}{k} \ln \left[ \frac{r_8-1}{r_8+1} \cdot \frac{r_1+1}{r_1-1} \right] \quad \text{for } L < 0 \quad (9a)$$

$$\text{where } r_8 = \left( 1 + 14 \times \frac{8}{|L|} \right)^{1/2}$$

$$r_1 = \left( 1 + 14 \times \frac{1}{|L|} \right)^{1/2}$$

or

$$\Delta\theta = \frac{T_*}{k} \left[ \ln 8 + 5.2 (8-1)/L \right] \approx T_* \left( 5.2 + \frac{q_1}{L} \right) \quad \text{for } L \geq 0 \quad (9b)$$

Thus for given  $u_*$  and  $T_*$ , equations (6) and (9) enable values of  $u(2)$  and  $\Delta\theta$  to be determined. It is clearly easier to go from  $u_*$  and  $T_*$  to  $u(2)$  and  $\Delta\theta$  than the reverse since  $L$  cannot readily be expressed in terms of  $u(2)$  and  $\Delta\theta$ . Whilst this is fine for the construction of graphs it has to be recognised that it is less convenient for numerical determinations of  $v_d$  from the measured variables  $u(2)$  and  $\Delta\theta$ . Here some form of iteration is required in which informed guesses for  $u_*$  and  $T_*$  are made, these are used to form  $L$  and then  $L$  is used to derive better estimates of  $u_*$  and  $T_*$  in terms of  $u(2)$  and  $\Delta\theta$ , a revised value of  $L$ , even better estimates of  $u_*$  and  $T_*$ , and so on until steady values are reached.

No matter the magnitude of  $H$ , for small enough  $z$  the  $\phi \approx 1$ . Thus if  $\phi_{\text{gas}} \equiv \phi_H$ ,  $\phi_{\text{gas}} = 1$  for small enough  $z$ . If  $q(z)$  is the concentration of the gas at height  $z$ , if  $E$  is the downward flux of the gas and  $z_q$  is the equivalent of  $z_0$ , then

$$q(z) - q(0) = \frac{E}{k \rho u_*} \ln \frac{z}{z_q}$$

where  $E$  by definition is given by  $E = \rho v_d(z) [q(z) - q(0)]$

Thus  $\frac{u_*}{v_d(z)} = \frac{1}{k} \ln \frac{z}{z_q}$  for small enough  $z$ .

Inserting this into equation (1) and assuming for the same small  $z$  that  $\frac{\bar{u}(z)}{u_*} = \frac{1}{k} \ln \frac{z}{z_0}$  it follows that

$$B^{-1} = \frac{1}{k} \ln \frac{z_0}{z_q} \quad \text{independent of } z. \quad (10)$$

or

$$z_q = z_0 \exp(-k B^{-1}) = z_0 \exp(-a k u_*^{1/3}) \quad (11)$$

For more general  $z$ :

$$v_d(z) = \frac{k u_*}{\int_{z_q}^z \frac{1}{z} \phi_H \left( \frac{z}{|L|} \right) dz} \quad (12)$$

which by generalising equations (9) can be seen to take the forms:

$$v_d(z) = \frac{k u_*}{\ln \left[ \frac{r-1}{r+1} \cdot \frac{r_q+1}{r_q-1} \right]} \quad \text{for } L < 0 \quad (13a)$$

where

$$r = \left( 1 + 14 \frac{z}{|L|} \right)^{1/2}$$

$$r_q = \left( 1 + 14 \frac{z_q}{|L|} \right)^{1/2}$$

and

$$v_d(z) = \frac{k u_*}{\ln \frac{z}{z_q} + \frac{5.2}{L} (z - z_q)} \quad \text{for } L \geq 0 \quad (13b)$$

where  $z_q$  is given by equation (11).

## Results

For specified  $z_0$  and parameter  $a$ , equations (6), (9), (11) and (13) can be



solved to give  $u(2)$ ,  $\Delta\theta$ ,  $z_0$  and  $v_d(z)$  in terms of  $u_*$ ,  $T_*$  and  $z$ . The results are displayed in Figures (1) - (4). Figure (1) shows the effect of parameter  $a$  on  $v_d$  at 1 metre in moderately unstable conditions. Figure (2) shows the sensitivity of  $v_d$  at 1 metre to  $z_0$  and to the surface sensible heat flux  $H$  for a given geostrophic wind. If a polluted air mass passes over heterogeneous terrain in which  $z_0$  varies by two orders of magnitude, say, then  $v_d(1)$  might vary by a factor of 3 according to this Figure. The variation of the deposition per unit area would probably be less than this because  $q(1)$  would tend to decrease over areas of high  $v_d(1)$  and vice versa.

Figure (3) shows two rather typical  $v_d(z)$  profiles in unstable and stable conditions. In unstable conditions  $v_d$  varies only by some 20% above 1 metre, and most of this variation occurs below 10 metres. In stable conditions a much greater variation is seen, reflecting the larger variation in concentration due to stabilisation.

Figure (4) is a nomogram for estimating  $v_d$  at 1 metre in terms of  $u(2)$  and  $\Delta\theta$  for  $z_0 = 1$  cm. The parameter  $a$  is taken to be 10. In stable conditions  $v_d$  depends almost entirely on  $u(2)$  :

$$v_d \approx 2.8 u(2) \quad (u \text{ is in } m s^{-1}, v_d \text{ in } cm s^{-1}) \quad (14)$$

whereas in unstable conditions  $v_d$  increases significantly with increasing  $\Delta\theta$ . Contours of heat flux  $H$  and Pasquill Stability  $P$  are also shown (Pasquill, 1974). In near neutral and unstable conditions  $H$  depends principally on  $\Delta\theta$ , varying only slowly with  $u(2)$ , whereas in stable conditions ( $P > 5$ )  $H$  depends mainly on  $u(2)$  and hardly at all on  $\Delta\theta$ . On the stable side, two curves are given for each stability value of  $P$  to indicate the present uncertainty in the validity of a simple assumption that  $P = P(u, \Delta\theta)$ , and the range of values implied by very limited data (Smith, unpublished).

The important conclusion is one that has been known for a long time but is often "conveniently forgotten", namely that  $v_d$  is not only a function of height  $z$ , but also depends quite critically on the state of the surface layer and chiefly on  $u$  and  $\Delta\theta$ .

Nevertheless the picture is a little simpler when we consider the deposition rate of a gas being emitted from a source a long way upstream. Ignoring for a moment the response of the total depth of the cloud of gas to wind speed, there is a tendency for the increase in  $v_d$  as wind speed increases to be balanced by a corresponding decrease in local concentration due to dilution effects at source. The deposition rate per unit surface area is by definition the product of the two and therefore depends only weakly on the state of the air, provided this is virtually constant. If the depth of the cloud increases significantly with increasing wind speed then the deposition rate will decrease and more material will travel to longer range. However on many occasions, perhaps even the majority of occasions, vertical mixing will be controlled by factors other than wind speed (e.g. by an overhead inversion), and this means that as a first approximation it may be



sufficient in long range transport models to ignore the influence of  $u$  and on deposition rates unless "extreme" stability conditions (A and G) are encountered.

### Surface Resistance

Many gases experience a further resistance to their take-up at the surface. Sulphur dioxide is one of these. The extra resistance  $r_s$  reflects the apparent fact that such gases are absorbed most readily within the stomata on leaves, although other factors may also play a role. The magnitude of  $r_s$  and its dependence on the state of the soil and the vegetation has been broadly estimated in studies of the surface energy balance and the surface heat flux (Wang, forthcoming publication) and it may be argued that the same values should apply to gases like sulphur dioxide. Except at night, or within an hour of sunrise or sunset, the following rough values have been inferred from the energy balance studies:

		$r_s$ (s m <sup>-1</sup> )
Oct. - March :	soil contains moisture	88
	no rain for over 10 days	400
April- Sept. :	soil contains moisture	50
	soil dry	2000

The value of the dry deposition velocity  $v_d'$  for such gases is then given by:

$$\frac{1}{v_d'} = r_s + \frac{1}{v_d}$$

Data collected at Barton Aerodrome in 1973 by the CEEB (Sugden et al., 1976) will be used to test the applicability of such a correction. The important data from their Tables II and III are reproduced below together with inferences concerning the sensible heat flux  $H$ ,  $r_s$  (expressed in units of s cm<sup>-1</sup>), and the implied  $v_d'$ . Sugden et al.'s observed values of  $v_d'$  are deduced from measurements of SO<sub>2</sub> concentration at two levels. The accuracy of such estimates is probably not very high and some anomalies are apparent in the data : e.g. on June 12th, three consecutive estimates of  $v_d'$  in periods in which the meteorology and other relevant factors changed very little were 0.74 cm s<sup>-1</sup>, 2.29 and 1.02.

The "observed"  $v_d'$  have a mean value of 0.95 cm s<sup>-1</sup>, and the calculated  $v_d'$  have a mean 0.89 cm s<sup>-1</sup> (in reasonably good accord), whereas the uncorrected  $v_d$  have a mean 2.17 cm s<sup>-1</sup> which is much too high. The correlation coefficient between  $v_d'$  calc. and the  $v_d'$  obs. is 0.61, which in view of the uncertainty in the input data is quite reasonable. Figure 5a shows a plot of the calculated  $v_d'$  against the observed  $v_d'$ , showing that without the  $r_s$ -correction the agreement is quite poor, all the  $v_d'$  being significantly larger than the  $v_d$ . Figure 5b on the other hand shows better overall agreement, albeit still with quite a lot of scatter. Obviously with such limited data no claim can be made that the method has general validity, but at least the results are promising.



Table: BARTON data on SO<sub>2</sub> Deposition Velocities

Date	Julian day	No.	Time	Cloud	$u_1$	$u_2$ (approx)	$u_{10}$ (approx)	T	$T_{\text{dew pt.}}$	$r_s$	H (estim.)	$v_d$ calc	$v_d$ calc.	$v_d$ obs.
2.3.73	61		12	6	4.5	5.2	5.8	8.3	4	0.8	28	2.1	0.78	1.57
			13	5	4.9	5.7	6.4	9.8	4.5	0.8	12	2.3	0.81	1.42
			14	8	5.2	5.9	6.7	9.6	4	0.8	0	2.3	0.81	1.17
			15	8	4.2	4.8	5.4	9.5	5	0.8	0	1.9	0.75	0.50
13.3.73	72		12	8	3.4	3.9	4.4	6.2	-1	0.8	25	1.7	0.72	0.42
			14	8	2.3	2.6	3.0	6.4	1	0.8	25	1.3	0.64	0.27
			15	8	1.8	2.1	2.3	6.5	0	0.8	6	1.0	0.56	0.36
23.3.73	82		12	2	3.9	4.5	5.1	13.5	0	4.0	164	2.1	0.22	0.21
26.4.73	116		15	5	4.5	5.1	5.8	13	5	0.5	19	2.1	1.02	1.64
15.5.73	135		15	7	1.7	1.9	2.1	13.8	0	0.5	41	1.2	0.75	0.56
			16	7	1.3	1.5	1.6	14	3	0.5	28	1.0	0.67	0.46
17.5.73	137		14	7	7.8	9.0	10.1	17.4	-2	0.5	0	3.3	1.25	0.93
			15	6	7.9	9.1	10.2	17.1	-4	0.5	0	3.3	1.25	1.28
			16	7	8.7	10.0	11.3	15.5	-1	0.5	0	3.5	1.27	1.14
12.6.73	163		12	7	5.6	6.4	7.2	16.7	11	0.5	20	2.4	1.09	1.20
			13	7	5.7	6.6	7.4	16.9	11	0.5	14	2.5	1.11	0.74
			14	7	6.1	7.1	7.9	19.1	9	0.5	0	2.7	1.15	2.29
			15	7	5.5	6.4	7.1	17.7	11	0.5	0	2.4	1.09	1.02

Units: cloud in oktas (eighths),  $u$  in  $\text{m s}^{-1}$ ,  $T$  in  $^{\circ}\text{C}$ ,  $r_s$  in  $\text{s cm}^{-1}$ ,  $H$  in  $\text{W m}^{-2}$ ,  $v_d$  in  $\text{cm s}^{-1}$



## Conclusions

These may be summarised as follows:

- (i) the deposition velocity may be assessed using Monin-Obukhov similarity theory and the inverse Stanton number ( $B^{-1}$ ) concept used by Chamberlain (1966).
- (ii)  $B^{-1}$ , according to Thom (1972), can be written as  $au_*^{\frac{1}{3}}$  where  $a$  is virtually independent of the nature of the vegetated surface and depends only on the character of the gas. Parameter  $a$  has to be determined experimentally.
- (iii)  $v_d$  varies rather slowly with heat flux  $H$  and with the parameter  $a$ , but varies significantly with height  $z$ , especially in stable conditions.
- (iv) except in extreme stability conditions,  $v_d$  is almost proportional to wind speed whereas the deposition rate per unit area is often independent of  $u$  and  $\Delta\theta$  provided they are almost constant along the path of the cloud.
- (v) Many gases of practical importance (like sulphur dioxide) do not find vegetated surfaces a perfect sink, and an extra surface resistance  $r_s$  has to be incorporated. The simple scheme presented for estimating  $r_s$  considerably improves the agreement between calculated and limited observed deposition velocities.
- (vi) Over vegetated surfaces, the incorporation of  $r_s$  reduces the deposition velocity to values typically in the range  $0.5 - 1.4 \text{ cm s}^{-1}$ . This is very close to the range of values expected over sea surfaces (see Figure 2) where  $z_0 = 10^{-5} - 10^{-4}$  and  $r_s = 0$ . Thus  $v_d'$  appears to be rather insensitive to the nature of the underlying surface, at least those commonly encountered in and around Europe.

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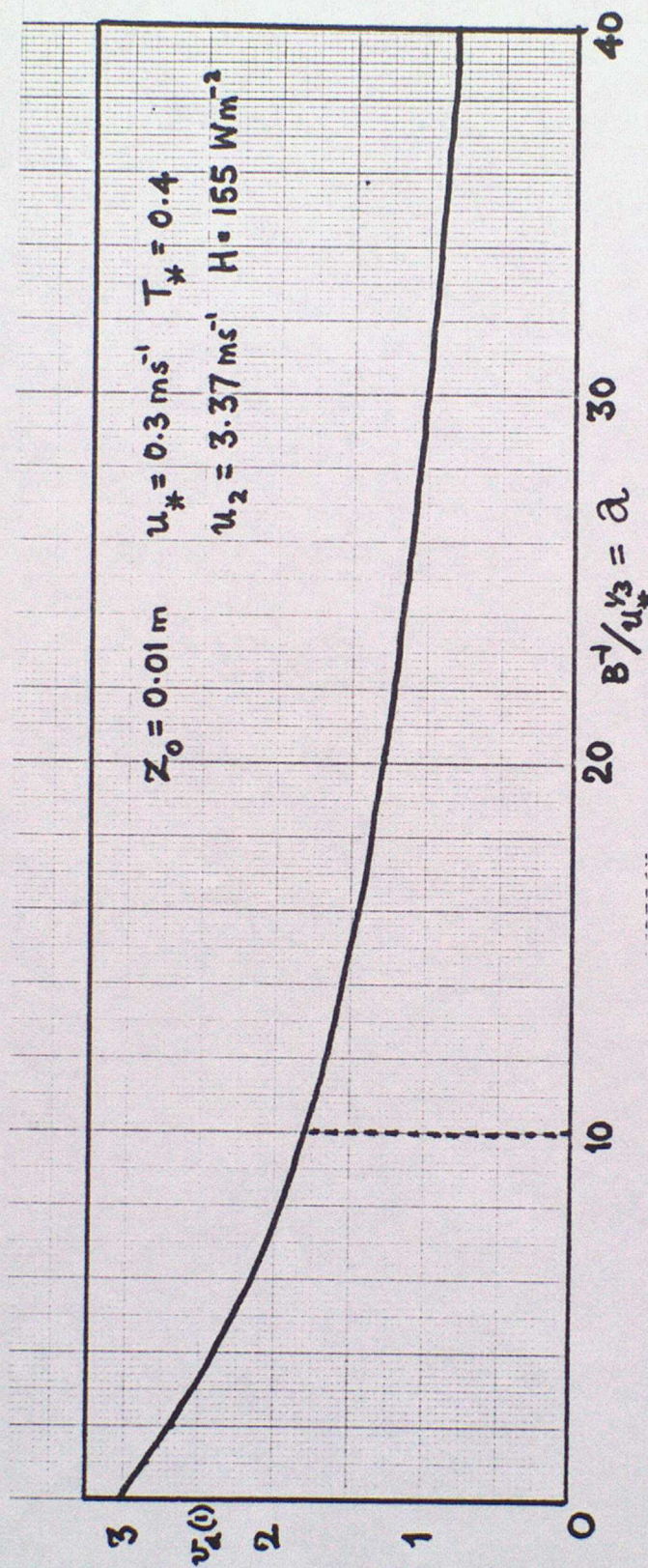


Figure 1. The variation of the deposition velocity  $v_d$  with the parameter  $a$  that appears in Thom's formulation in the inverse Stanton number  $B^{-1}$ .



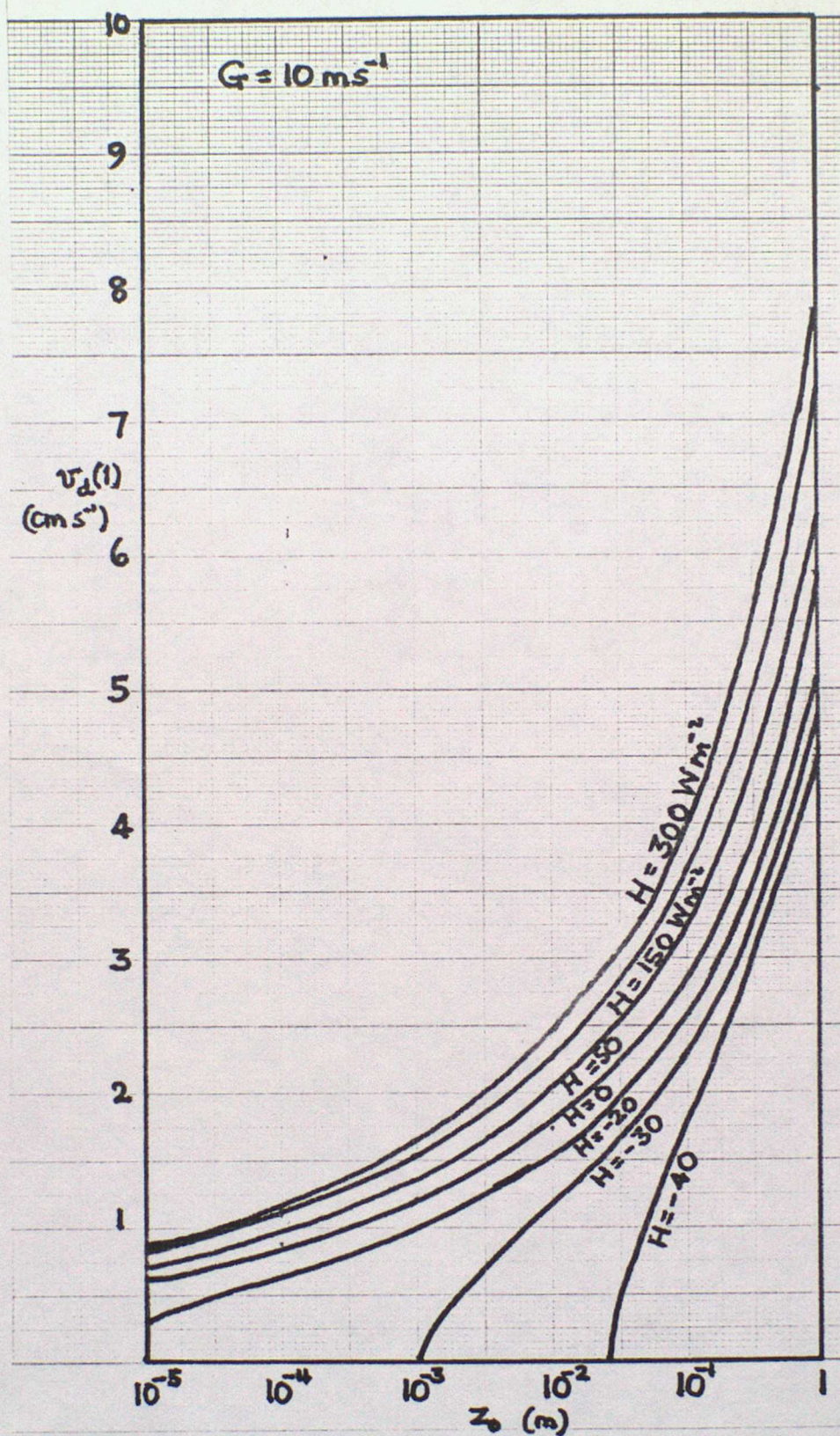
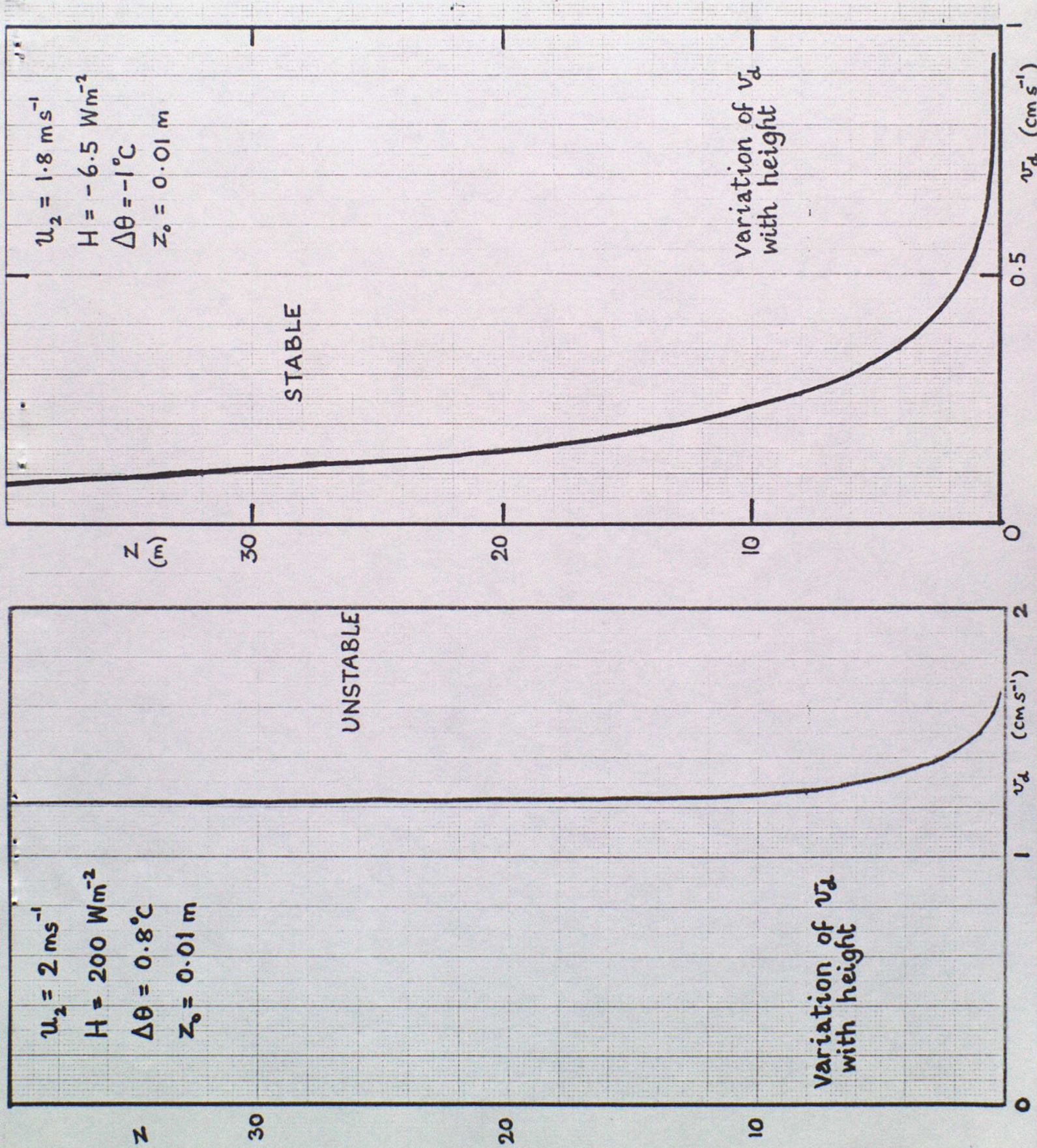


Figure 2. Variation in the deposition velocity  $v_d$  with surface roughness and heat flux  $H$  for a specified geostrophic wind speed  $G$ .



Figure 3.  
Typical profiles  
of  $v_d$  in unstable  
and stable conditions.





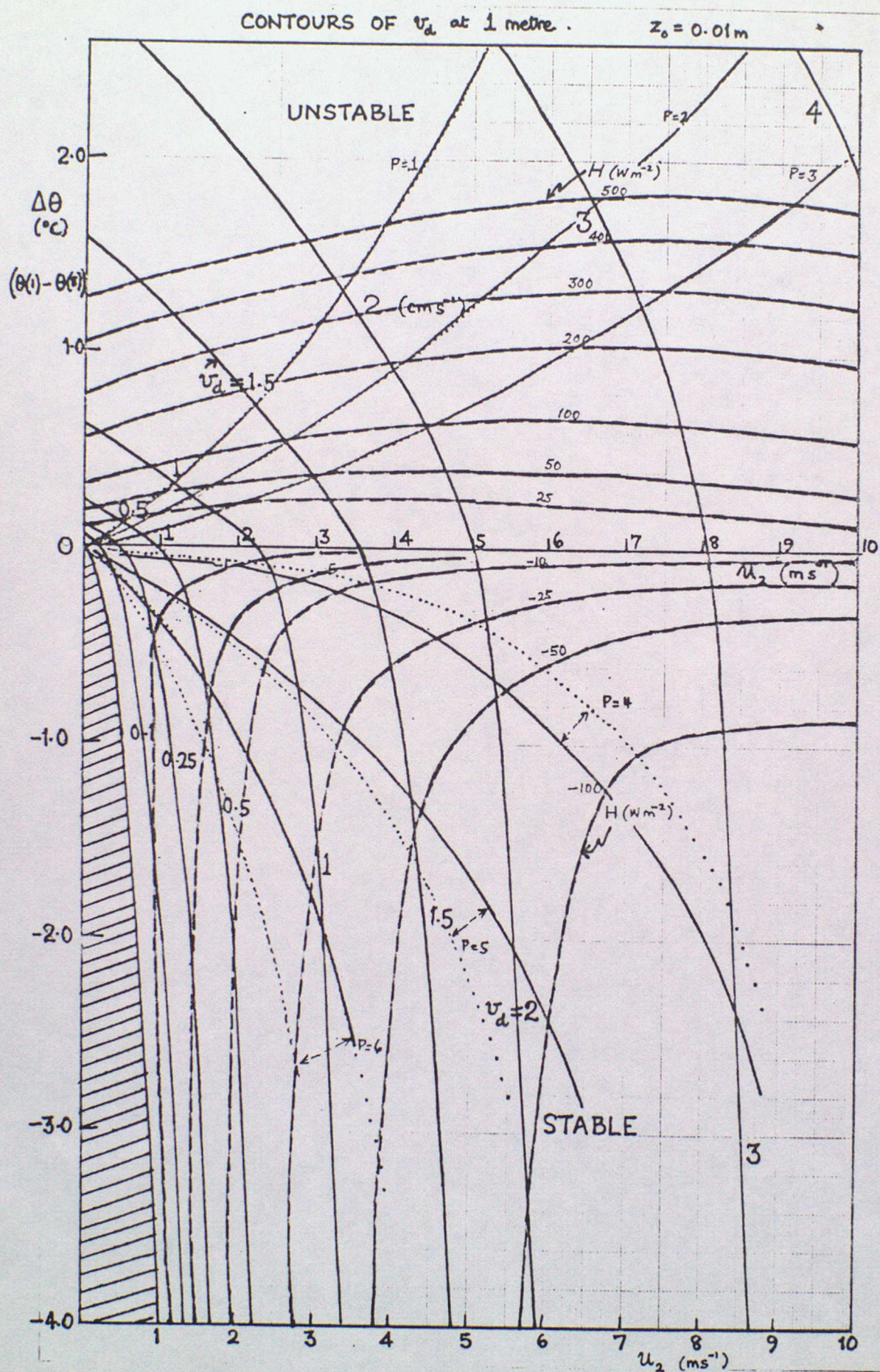


Figure 4. Contour lines of deposition velocity  $v_d$ , heat flux  $H$ , and Pasquill Stability  $P$ , for  $z_0 = 0.01m$  and parameter  $a = 10$ .



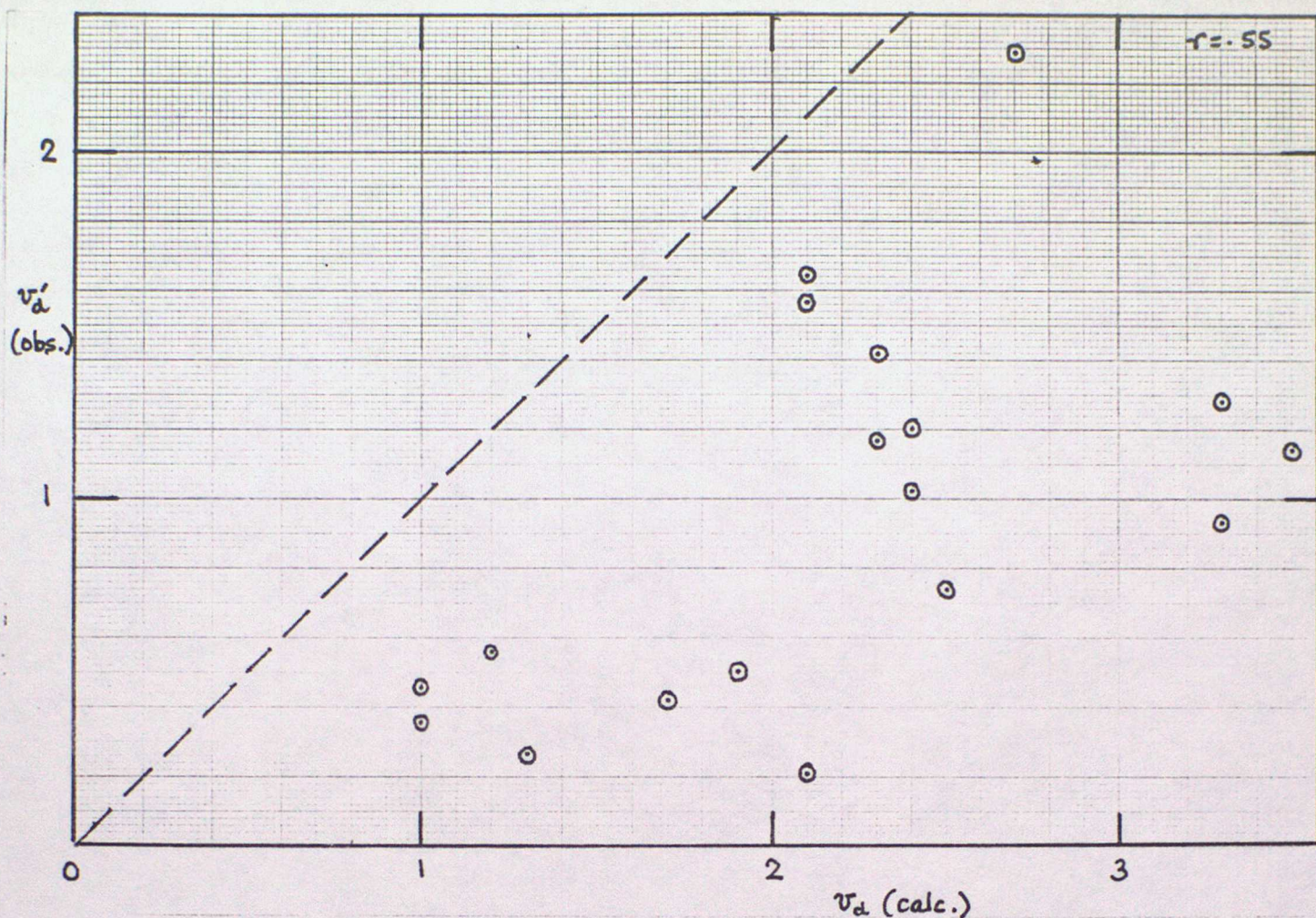


Figure 5a. A comparison of uncorrected  $v_d$  (calculated) and "observed" deposition velocities taken at Barton (Sugden et al. 1976). The calculated values are clearly too large.

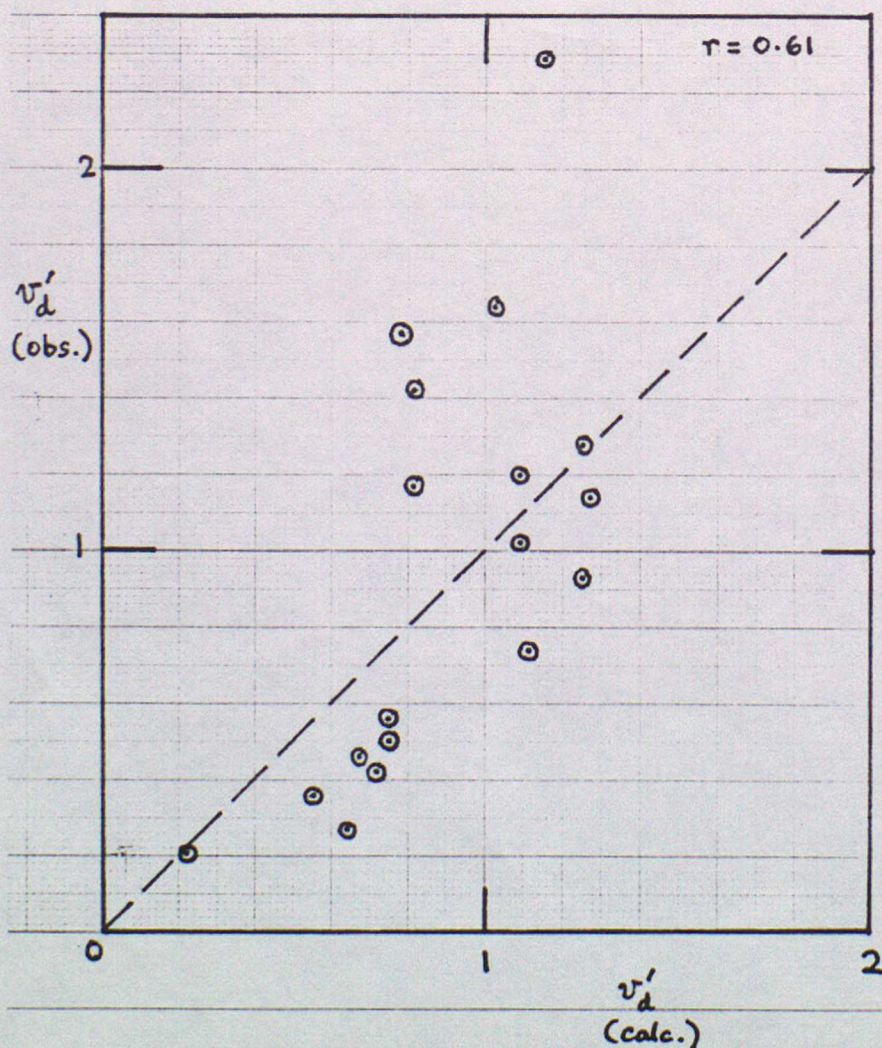


Figure 5b. The calculated values are now corrected by incorporation of the surface resistance  $r_s$ . Agreement with the "observed" is considerably improved although errors in the input data still give rise to significant scatter.